
Another possible origin of temperature and pressure gradients across vanes in the Crookes radiometer

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The Crookes Radiometer [1,2]

- 4 vanes in a glass bulb partially evacuated.
- One side of vane is black and the other side is shiny.
- Vanes revolve with shiny side leading under sunlight.



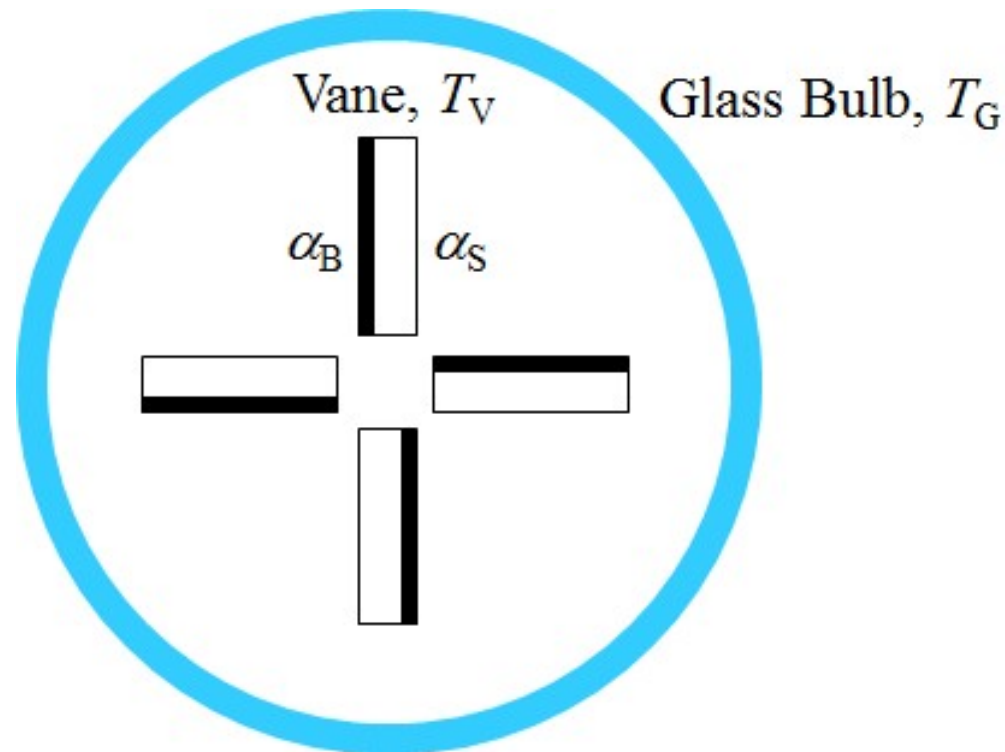
Past Simulation Studies [3-10]

- Great efforts made by many researchers to reveal forces on vanes
 - thermal transpiration / thermal creep force due to ΔT
 - area force by Δp
- Assumptions used in every work
 - temperature at black side of vane is higher than that at the shiny side, $T_B > T_S$.
 - accommodation coefficient α is uniform and same at both sides of vane.



New Hypothesis proposed in This Study

- Vanes is **isothermal at T_V** .
- Accommodation coefficient α_B at black side of vane is different from that at shiny side α_S , and **$\alpha_B > \alpha_S$** .



Estimating Vane Temperature

- Heat balance equations under Biot number $Bi \ll 1$

$$\begin{cases} q_{in} - (q_{g,B}^t + q_{r,B}^t) = -\kappa \frac{\partial T_V^t}{\partial x} \\ \rho L_b C_p \frac{\partial T_V^{t+\Delta t}}{\partial t} = q_{in} - (q_{g,B}^t + q_{r,B}^t) - (q_{g,S}^t + q_{r,S}^t) \end{cases}$$

$$q_{g,B/S}^t = \frac{1}{4} n \bar{v} \Delta E = \frac{1}{2} nk \sqrt{\frac{8kT_g}{\pi m}} (T_{V,B/S}^t - T_g)$$

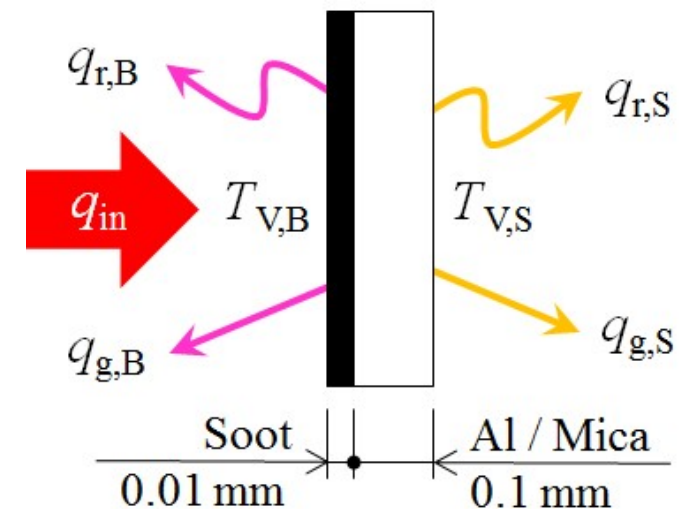
$$q_{r,B/S}^t = \varepsilon_{B/S} \sigma \left\{ (T_{V,B/S}^t)^4 - T_g^4 \right\}$$

Ambient

Air, 1 Pa
 $T_g = 298$ K

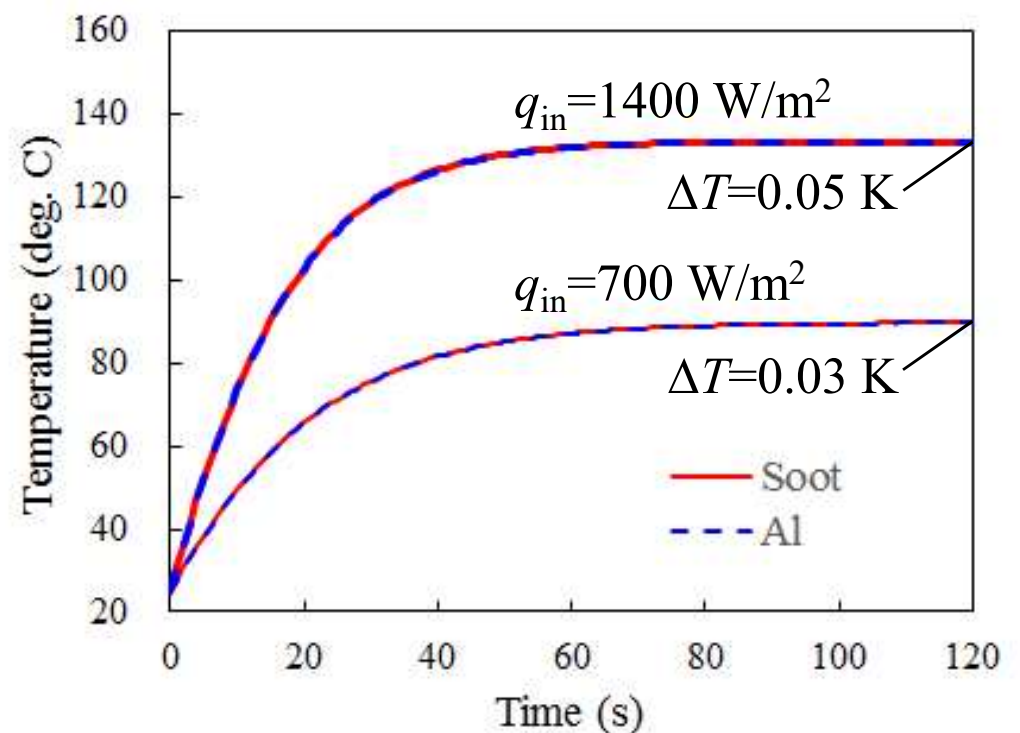
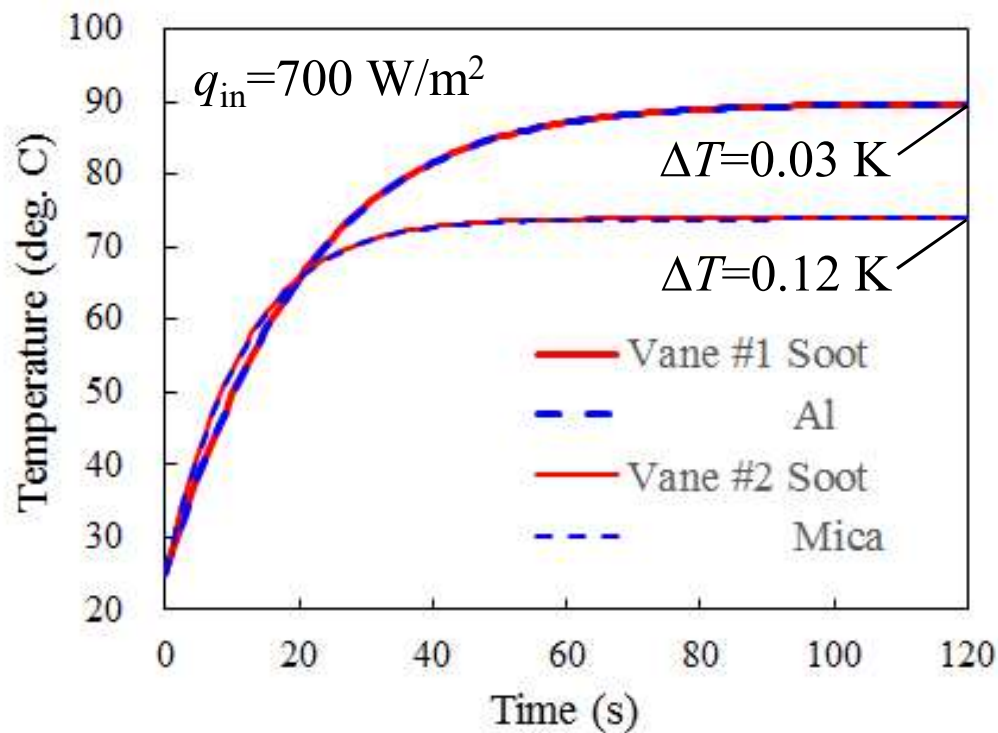
Material Properties [11-14]

	ρ (kg/m ³)	C_p (J/kg-K)	κ (W/m-K)	ε
Al	2688	905	237	0.17
Mica	2100	880	0.5	0.72
Soot	100	1000	0.05	0.95



Estimating Vane Temperature (cont'd)

- Typical heat flux of sunlight is 700 – 1400 W/m² [15,16]
- Calculated Biot number $Bi < 0.01$.
- **Vane is isothermal** under sunlight.

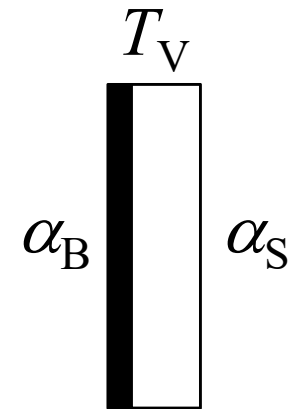




DSMC_2D.xls

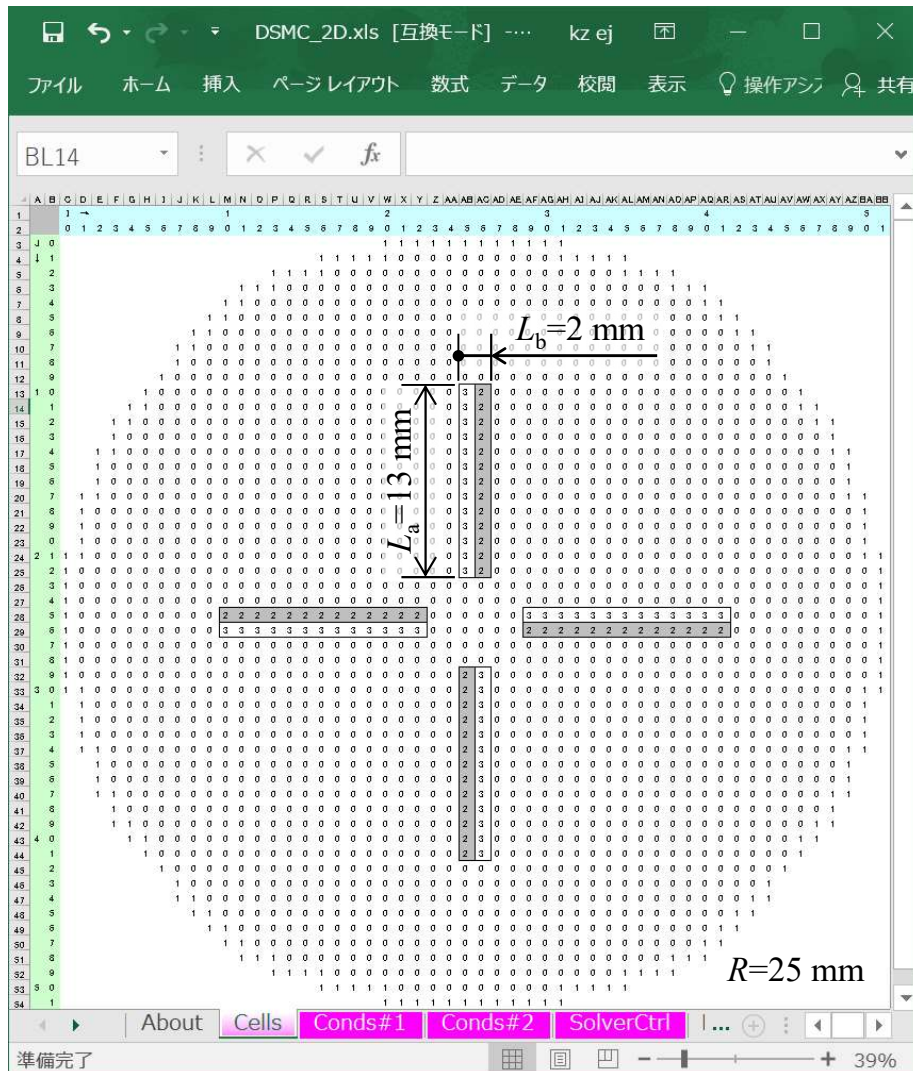
DSMC_2D.xls [17]

- Multipurpose 2D DSMC software created on MS-Excel
 - www2b.biglobe.ne.jp/~denpoh/Software/DSMC_xls/
- Gas (Air)
 - Diatomic molecule with rotational degrees of freedom
 - Molecular model: Maxwell molecule
 - Collision models: VHS model, Larsen-Borgnakke model
- Accommodation coefficients
 - Black side: $\alpha_B = 1$ (diffuse reflection)
 - Shiny side: diffuse reflection α_S + specular reflection $(1 - \alpha_S)$



Model Setup in DSMC_2D.xls

- Vane length $L_a = 13$ mm, thickness $L_b = 2$ mm



	A	B	C	D	E
1			Value	Unit	Note
2					
3	Cells	dx	1.00E-03	m	Cell size in I-direction
4		dy	1.00E-03	m	Cell size in J-direction
5					
6	Gas Properties	Molecular Weight	28.970	g/mol	
7		Viscosity	1.94210E-05	Pa*s	@Tref
8		Internal Degree of Freedom	2		IDF = Int(2*Cp/R-5)
9					
10	Reference Values	Pressure	1.000E+00	Pa	
11		Temperature	322.031	K	
12					
13	Initial Conditions	Pressure	1.000E+00	Pa	
14		Temperature	322.031	K	
15		# of Super Particles per Cell	50		
16					
17	Walls	# of Walls	3		Max = 9
18					
19	Upstream BCs	Wall #	8		
20		Type	1		1-Pressure, 2-Velocity
21		Pressure	1.000E+00	Pa	
22		Velocity	0.000	m/s	
23					
24	Downstream BCs	Wall #	9		
25		Type	1		1-Pressure, 2-Reflection Probability, 3-Perfect Vacuum
26		Pressure	1.000E-01	Pa	< Upstream Pressure

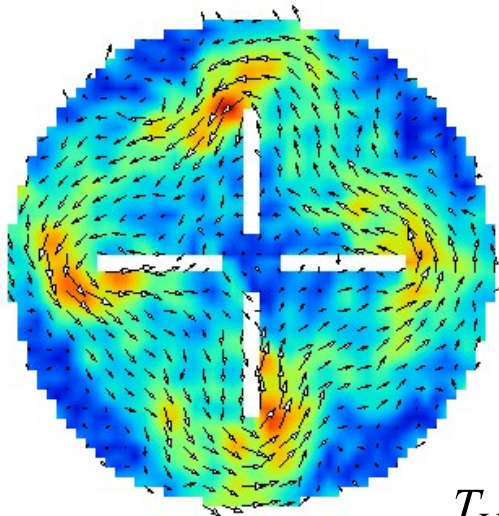
	A	B	C	D	E	F	G
1							
2			Probability				
3	Wall #	Temperature (K)	Diffuse Reflection	Specular Reflection	Sticking	TOTAL	CHECK
4	1	298.000	1.000			1.000	
5	2	348.000	1.000			1.000	
6	3	348.000	0.010	0.990		1.000	
7	4						
8	5						
9	6						
10	7						
11	8						
12	9						



Example Flow Fields

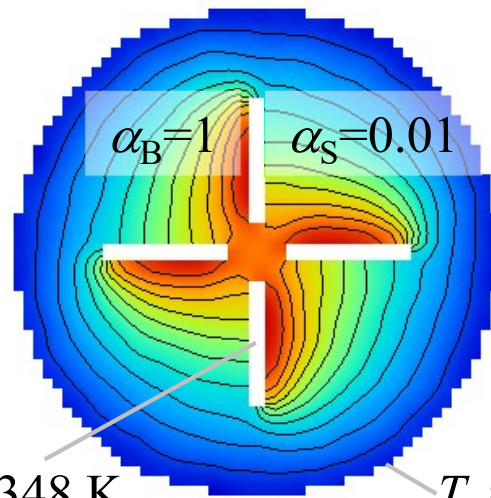
- ΔT and Δp across vane are produced.
- ΔT induces thermal creep flow.
- Δp acts as area force to push vanes from black side.

Velocity (m/s)



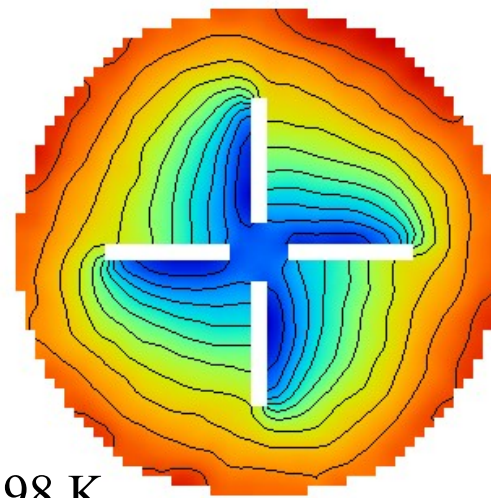
Min-Max = 1.5146

Temperature (K)



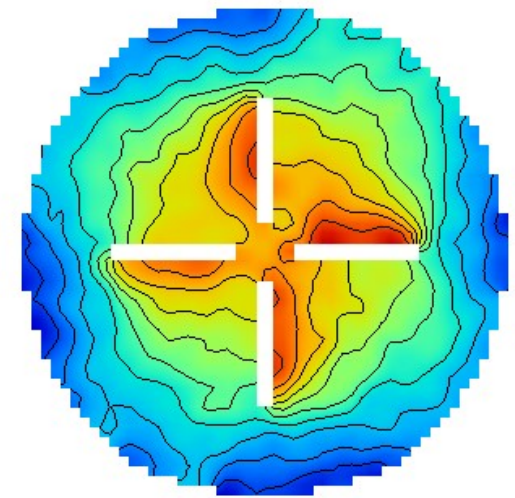
28.25

Density (m⁻³)



1.6530E+19

Pressure (Pa)

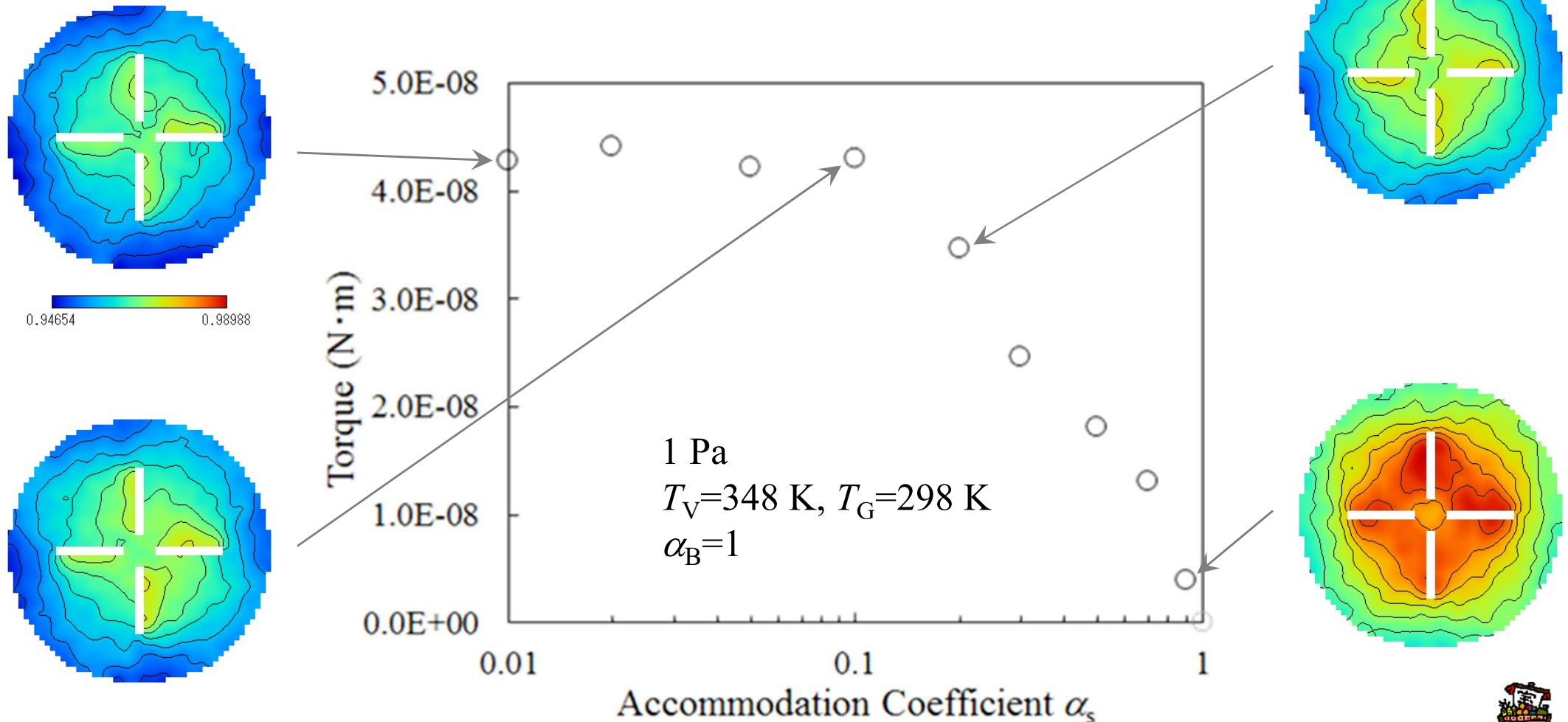


0.02625



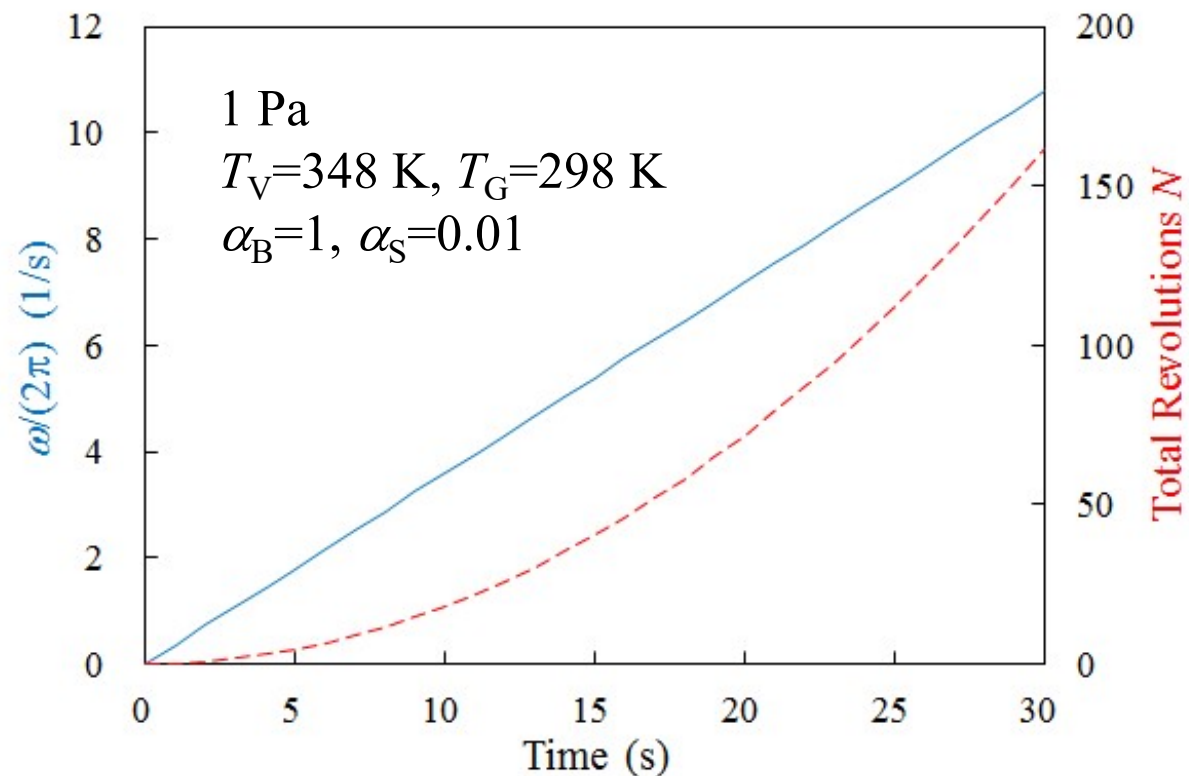
$\alpha_B > \alpha_S$ produces Torque

- Torque by Δp increases with decreasing α_S for $\alpha_S > 0.1$,
- then saturates for $\alpha_S < 0.1$.



Rotation Speed of Vanes

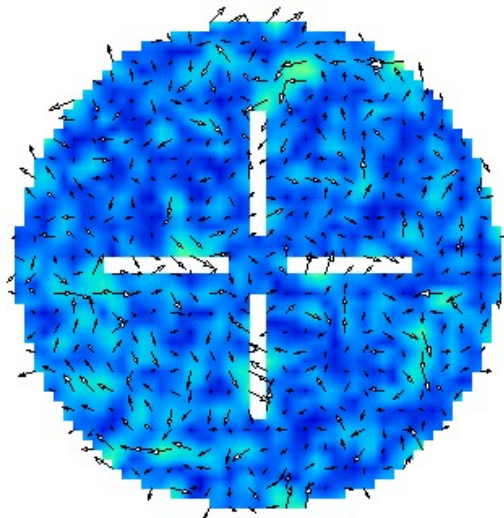
- Estimated by assuming torque of stationary vanes is the same as freely rotating vanes.
- Should be valid only at early state of starting rotation. [8]
- Time scale is sec-order as commonly observed.



What if Glass Bulb is Heated Up? ($T_G = T_V$)

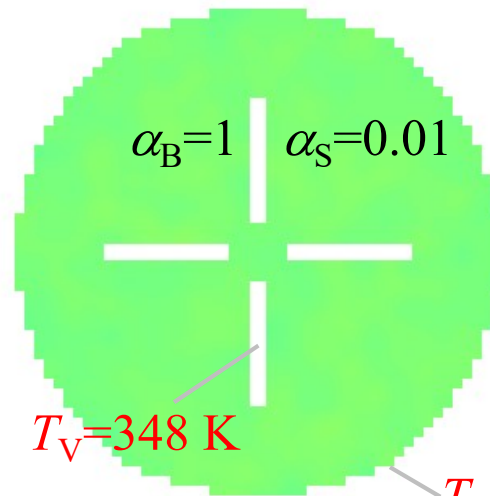
- Flow fields are uniform ($\Delta T \rightarrow 0$, $\Delta p \rightarrow 0$) even for $\alpha_B \gg \alpha_S$.
- Apparent thermal creep flow is not induced.
- Revolution of vanes will stop.

Velocity (m/s)



0 1.5146

Temperature (K)

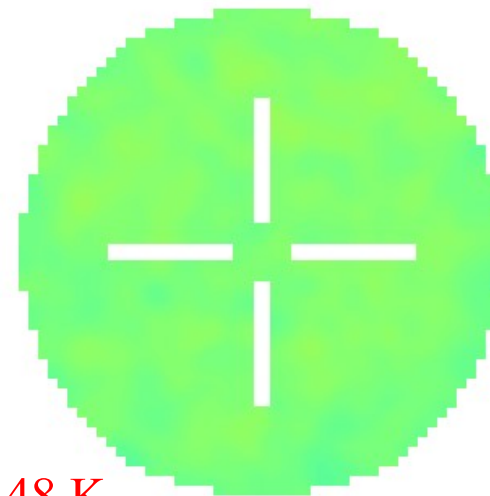


$T_V=348$ K

$T_G=348$ K

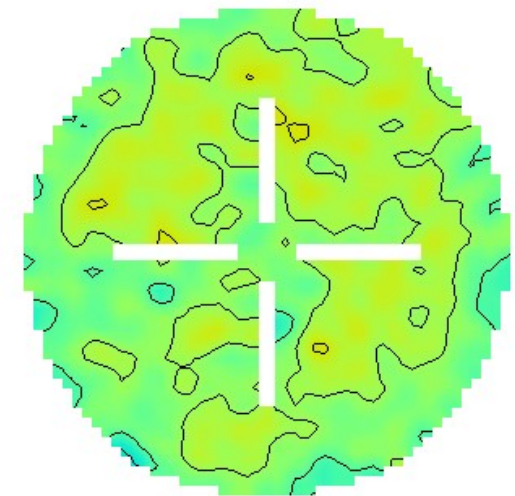
330.665 358.915

Density (m^{-3})



$1.9976\text{E}+20$ $2.1629\text{E}+20$

Pressure (Pa)



0.97696 1.00321



Summary

- New hypothesis
“Vane is isothermal, and $\alpha_B > \alpha_S$ ”
has been proposed and investigated using heat transfer and DSMC simulations.
- The results have proved
 - vane is isothermal under sunlight, and
 - contrast of α_B and α_S can be an origin of ΔT and Δp across vane.
 - Δp works as an area force to push vanes.
- Also found glass bulb temperature strongly affects revolution of vanes.



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